Seat No.: _____

GUJARAT TECHNOLOGICAL UNIVERSITY BE - SEMESTER-III EXAMINATION – WINTER 2015

	•	Code:133502 Date:21/12/2015	
	-	Name: Analytical Techniques :30pm to 5:00pm Total Marks: 70	
Ins	2.	Attempt all questions.	
Q.1	(a)	Define the term: Chromatography. Explain theory and instrumentation of Gas chromatography.	07
	(b)	What are titrations? Explain EDTA titration in detail with procedure and calculation.	07
Q.2	(a)	i. Write a note on most widely used solid stationary phase in GC.	04
		ii. Write a short note on sampling loop used in chromatography.	03
	(b)	Explain principle, instrumentation and working of IR spectroscopy.	07
		OR	
	(b)	What is good laboratory practises (GLP)? Explain in detail.	07
Q.3	(a)	Explain volumetric estimation of brass alloy with procedure and calculation.	07
	(b)	What are the characteristics should pumps have used in HPLC. Discuss any one in detail with diagram.	07
		OR	
Q.3	(a)	Write a note on column Chromatography.	07
	(b)	Describe in detail the instrumentation for scanning the mass spectrum of an organic compound.	07
Q.4	(a)	i. What is Rf value? Explain various factors affecting in TLC.	04
		ii. Write a short note on guard column used in HPLC.	03
	(b)	Analysis of sample gave following values of Cu content: 41.20, 41.33, 41.60, 41.37 and 41.27.Calculate the mean, median, standard deviation, coefficient of	07

variance and range.

OR

Q.4	(a)	Define the term: co-precipitation and post precipitation. Explain Gravimetric estimation of Ni.	07
	(b)	Write notes on Chemical shift, shielding, deshielding effect and spin-spin coupling.	07
Q.5	(a)	Explain different types of transitions involved in UV-Visible spectroscopy with examples.	07
	(b)	Write a short note on Finger print region. How will you distinguish Ethanol and dimethyl ether using IR spectrum? OR	07
Q.5	(a)	Explain Lambert Beer's law of absorption with derivation.	07
	(b)	An organic compound (molecular formula $:C_8H_8O$) exhibits the following spectral data:	07
		IR: $3042 \text{ cm}^{-1}(\text{m})$, $2862 \text{ cm}^{-1}(\text{w})$, $2740 \text{ cm}^{-1}(\text{ w})$, $1722 \text{ cm}^{-1}(\text{s})$, 1605 cm^{-1} and 1575 cm^{-1}	
		UV: λ_{max} at 292 nm	
		NMR: δ 7.28 (5H, multiplet), 2.8 (2H,doublet), 9.78 (1H,triplet)	
		Deduce the structure of the compound.	