## Maximum: 100 marks

Time: 1 hour and 15 minutes

(A) 0.1 M (B) 1 M (C) 0.5 M (D) 10 M  2. In the reaction N <sub>2</sub> O <sub>4(g)</sub> \( \infty \) 2 NO <sub>2(g)</sub> , an increase in pressure:  (A) Terminate the reaction (B) Shifts the equilibrium to the left (C) Does not change the equilibrium (D) Shifts the equilibrium to the right  3. A colligative property depends on:  (A) The number of solvent particles in which the solute dissolves (B) Mass of solute dissolved in the solvent (C) Mass of solvent taken  (D) The number of solute particles dissolved in a given volume of the solvent  4. On dilution, specific conductance of an electrolyte:  (A) Increases (B) Decreases (C) Does not change (D) Increases and then decreases  5. pH of a solution whose hydroxide ion concentration is 0.001 mole/litre will be:  (A) 3 (B) 11  (C) 12 (D) 2  6. The conjugate acid-base pairs in the following reaction, HCl + NH <sub>3</sub> \(\infty \) NH <sub>4</sub> + Cl <sup>-</sup> , are:  (A) HCl and Cl <sup>-</sup> (B) NH <sub>3</sub> and Cl <sup>-</sup> (C) HCl and NH <sub>3</sub> (D) NH <sub>4</sub> and Cl <sup>-</sup>	1.	The mola	rity of the solution containing 4 g of s	odium l	ydroxide in one litrewater is:
2. In the reaction N <sub>2</sub> O <sub>4(g)</sub> \( \simes \) 2 NO <sub>2(g)</sub> , an increase in pressure:  (A) Terminate the reaction (B) Shifts the equilibrium to the left (C) Does not change the equilibrium (D) Shifts the equilibrium to the right  3. A colligative property depends on:  (A) The number of solvent particles in which the solute dissolves (B) Mass of solvent taken (C) Mass of solvent taken (D) The number of solute particles dissolved in a given volume of the solvent  4. On dilution, specific conductance of an electrolyte:  (A) Increases (B) Decreases (C) Does not change (D) Increases and then decreases  5. pH of a solution whose hydroxide ion concentration is 0.001 mole/litre will be:  (A) 3 (B) 11 (C) 12 (D) 2  6. The conjugate acid-base pairs in the following reaction, HCl + NH <sub>3</sub> \(\simes \) NH <sub>4</sub> + Cl <sup>-</sup> , are:  (A) HCl and Cl <sup>-</sup> (B) NH <sub>3</sub> and Cl <sup>-</sup> (C) HCl and NH <sub>3</sub> (D) NH <sub>4</sub> and Cl <sup>-</sup> 7. The number of fundamental vibrations of methane are:  (A) 3 (B) 4		000000		Townson Co.	
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(C) Mass of solvent taken (D) The number of solute particles dissolved in a given volume of the solvent  4. On dilution, specific conductance of an electrolyte:  (A) Increases (B) Decreases (C) Does not change (D) Increases and then decreases  5. pH of a solution whose hydroxide ion concentration is 0.001 mole/litre will be:  (A) 3 (B) 11 (C) 12 (D) 2  6. The conjugate acid-base pairs in the following reaction, HCl + NH <sub>3</sub> NH <sub>4</sub> + Cl <sup>-</sup> , are:  (A) HCl and Cl <sup>-</sup> (B) NH <sub>3</sub> and Cl <sup>-</sup> (C) HCl and NH <sub>3</sub> (D) NH <sub>4</sub> and Cl <sup>-</sup> 7. The number of fundamental vibrations of methane are:  (A) 3 (B) 4		(A)	The number of solvent particles in v	which th	ne solute dissolves
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(A) Increases (C) Does not change (D) Increases and then decreases  5. pH of a solution whose hydroxide ion concentration is 0.001 mole/litre will be:  (A) 3 (B) 11 (C) 12 (D) 2  6. The conjugate acid-base pairs in the following reaction, HCl+NH₃⇒NH₄+Cl⁻, are:  (A) HCl and Cl⁻ (B) NH₃ and Cl⁻ (C) HCl and NH₃ (D) NH₄+ and Cl⁻  7. The number of fundamental vibrations of methane are:  (A) 3 (B) 4		(D)	The number of solute particles disse	olved in	a given volume of the solvent
(C) Does not change  (D) Increases and then decreases  pH of a solution whose hydroxide ion concentration is 0.001 mole/litre will be:  (A) 3 (B) 11 (C) 12 (D) 2  The conjugate acid-base pairs in the following reaction, HCl+NH <sub>3</sub> NH <sub>4</sub> +Cl <sup>-</sup> , are:  (A) HCl and Cl <sup>-</sup> (B) NH <sub>3</sub> and Cl <sup>-</sup> (C) HCl and NH <sub>3</sub> (D) NH <sub>4</sub> and Cl <sup>-</sup> The number of fundamental vibrations of methane are:  (A) 3 (B) 4	4.	On dilutio	on, specific conductance of an electrol	yte:	
pH of a solution whose hydroxide ion concentration is 0.001 mole/litre will be:  (A) 3 (B) 11 (C) 12 (D) 2  6. The conjugate acid-base pairs in the following reaction, HCl + NH <sub>3</sub> NH <sub>4</sub> + Cl <sup>-</sup> , are:  (A) HCl and Cl <sup>-</sup> (B) NH <sub>3</sub> and Cl <sup>-</sup> (C) HCl and NH <sub>3</sub> (D) NH <sub>4</sub> and Cl <sup>-</sup> 7. The number of fundamental vibrations of methane are:  (A) 3 (B) 4		(A)	Increases	(B)	Decreases
(A) 3 (B) 11 (C) 12 (D) 2  6. The conjugate acid-base pairs in the following reaction, $HCl + NH_3 \rightleftharpoons NH_4^+ + Cl^-$ , are:  (A) $HCl$ and $Cl^-$ (B) $NH_3$ and $Cl^-$ (C) $HCl$ and $NH_3$ (D) $NH_4^+$ and $Cl^-$ 7. The number of fundamental vibrations of methane are:  (A) 3 (B) 11 (D) 2		(C)	Does not change	(D)	Increases and then decreases
(C) 12 (D) 2  6. The conjugate acid-base pairs in the following reaction, HCl + NH <sub>3</sub> NH <sub>4</sub> <sup>+</sup> + Cl <sup>-</sup> , are:  (A) HCl and Cl <sup>-</sup> (B) NH <sub>3</sub> and Cl <sup>-</sup> (C) HCl and NH <sub>3</sub> (D) NH <sub>4</sub> <sup>+</sup> and Cl <sup>-</sup> 7. The number of fundamental vibrations of methane are:  (A) 3 (B) 4	5.	pH of a so	olution whose hydroxide ion concentra	ation is	0.001 mole/litre will be :
6. The conjugate acid-base pairs in the following reaction, HCl+NH <sub>3</sub> NH <sub>4</sub> +Cl <sup>-</sup> , are:  (A) HCl and Cl <sup>-</sup> (B) NH <sub>3</sub> and Cl <sup>-</sup> (C) HCl and NH <sub>3</sub> (D) NH <sub>4</sub> and Cl <sup>-</sup> 7. The number of fundamental vibrations of methane are:  (A) 3 (B) 4		(A)	3	(B)	11
(A) HCl and Cl <sup>-</sup> (C) HCl and NH <sub>3</sub> (D) NH <sub>4</sub> and Cl <sup>-</sup> (D) NH <sub>4</sub> and Cl <sup>-</sup> (E) The number of fundamental vibrations of methane are: (A) 3 (B) 4		(C)	12	(D)	2
(A) HCl and Cl <sup>-</sup> (C) HCl and NH <sub>3</sub> (D) NH <sub>4</sub> and Cl <sup>-</sup> (D) The number of fundamental vibrations of methane are: (A) 3 (B) 4	6.	The conju	gate acid-base pairs in the following	reaction	1, $HCl + NH_3 \longrightarrow NH_4^+ + Cl^-$ , are:
7. The number of fundamental vibrations of methane are: (A) 3 (B) 4					
(A) 3 (B) 4		(C)	HCl and NH <sub>3</sub>	(D)	NH₄ and Cl⁻
(A) 3 (B) 4	7.	The num	ber of fundamental vibrations of meth	nane are	):
가게 하고 있는 것들은 사람들이 되었다. 나는 사람들이 없어 있는데 그렇게 되었다. 그 사람들이 되었다면 보고 있는데 그리고 있다면 보다 되었다. 그리고 있다면 보다 보다 되었다. 그리고 있다면 다른					
			9	(D)	8

8.	The symmetric vibrations of CO <sub>2</sub> are:				
	(A)	Raman active	(B)	IR active	
	(C)	Both Raman and IR active	(D)	Both Raman and IR inactive	
9.	An adsort	pent used in column chromatogra	phy is :		
	(A)	Sodium hydroxide	(B)	Silicon dioxide	
	(C)	Calcium hydroxide	(D)	Aluminum chloride	
10.	Which of	the following is an extensive prop	erty?		
	(A)	Pressure	(B)	Temperature	
	(C)	Concentration	(D)	Heat	
11.	pH of mil	k ranges from :			
	(A)	6-7	(B)	5-6	
	(C)	7-8	(D)	3-4	
12.	The yellow	w colour of milk is mainly due to :			
	(A)	Casein	(B)	Carotene	
	(C)	Riboflavin	(D)	Phospholipids	
13.	The prese	ence of starch in milk can be detec	ted using:		
	(A)	Bromine water	(B)	Calcium chloride solution	
	(C)	Iodine solution	(D)	Hydrochloric acid	
14.	An examp	ole for a synthetic food colour addi	itive :		
	(A)	Saffron	(B)	Curcumin	
	(C)	Caramel	(D)	Tartrazine	
15.	The titrab	ole acidity of milk can be expresse	d in terms	of:	
	(A)	Citric acid	(B)	Malic acid	
	(C)	Lactic acid	(D)	Tartaric acid	
16.	An examp	ele of a class II preservative is :			
	(A)	Honey	(B)	Acetic acid	
	(C)	Glucose	(D)	Benzoic acid	

17.	The presence of tin in the food sample can be detected using:						
	(A) Spectrophotometric Catechol violet method						
	(B)	B) Colorimetric Carbamate method					
	(C)	Colorimetric molybdenum	blue method				
	(D)	Colorimetric Silver diethyl	dithiocarbamate	method			
18.	An examp	le for an artificial sweetener					
	(A)	Sucrose	(B)	Polyphenols			
	(C)	Cyclamate	(D)	Fructose			
19.	Fat percer	nt of toned milk in Kerala :					
	. (A)	3.5	(B)	4			
	(C)	4.5	(D)	3			
20.	The test f	or checking the efficiency of	pasteurization of	f milk :			
	(A)	Alkaline phosphatase test	(B)	Gerbers test			
	(C)	CLR test	(D)	Mohrs method			
21.	The swee	test sugar among the followi	ng is:				
	(A)	Glucose	(B)	Mannose			
	(C)	Galactose	(D)	Fructose			
22.	First enzy	me to be isolated in pure cr	ystalline form wa	as:			
	11	Zymase	(B)	Invertase			
	(C)	Urease	(D)	Lactase			
23.	Hexokina	se enzyme belongs to the cla	iss:				
	(A)	Oxido-reductase	(B)	Transferase			
	(C)	Ligase	(D)	Hydrolase			
24.	Which of	the following is NOT a home	opolysaccharide?				
	(A)	Pectin	(B)	Chitin			
	(C)	Inulin	(D)	Cellulose			
25.	Caesin of	milk is a:					
	(A)	Glycoprotein	(B)	Phosphoprotein			
	(C)	Metalloporotein	(D)	Lipoprotein			

26.	Which of the following amino acid does not show optical activity?						
	(A)	Proline	(B)	Histidine			
	(C)	Valine	(D)	Glycine			
27.	Palmitie	acid is:					
	(A)	Monoenoic fatty acid	(B)	Polyenoic fatty acid			
	(C)	Unsaturated fatty acid	(D)	Saturated fatty acid			
28.	Proteins	absorbs UV radiation maximum at 2	80 nm d	ue to the presence of:			
	(A)	Tryptophan	(B)	Tyrosine			
	(C)	Phenyl alanine	(D)	All of the above			
29.	Two prote them will		l same is	soelectric point. The best way to resolve			
	(A)	Ion Exchange Chromatography	(B)	Gel Filtration Chromatography			
	(C)	Reverse Phase Chromatography	(D)	Chromatofocussing			
30.	Dietary fa	at:					
	(A) is usually present, although there is no specific need for it						
	<ul> <li>(B) if present in excess, can be stored as either glycogen or adipose tissue triacylglycerol</li> </ul>						
	(C)	(C) should include linoleic and linolenic acids					
	(D)	should increase on an endurance to energy stores	raining <sub>l</sub>	program in order to increase the body's			
31.	The effect	ts of vitamin A may include all of the	followin	ng EXCEPT:			
	(A)	Prevention of anemia	(B)	Serving as an antioxidant			
	(C)	Cell differentiation	(D)	The visual cycle			
32.	All the fol	llowing compounds are Intermediate	of the ci	tric acid cycle except :			
	(A)	Isocitrate	(B)	Malate			
	(C)	Pyruvate	(D)	Succinate			
33.	Which of	the following would probably not be a	affected	when a protein is denatured?			
	(A)	Primary structure	(B)	Secondary structure			
	(C)	Hydrogen bonds	(D)	Tertiary structure			

34.	The general name for an enzyme that transfers phosphate groups from ATP to a Protein is:			
	(A)	Protein kinase	(B)	Phosphorylase
	(C)	Phosphatase	(D)	ATPase
35.	A techniqu	ie commonly used for the assay of ho	rmones	is:
	(A)	Gel filtration	(B)	GLC
	(C)	Radioimmunoassay	(D)	HPLC
36.	Iodine nur	mber is an indication of:		
	(A)	Chain length	(B)	Degree of unsaturation
	(C)	Rancidity	(D)	Specific gravity
37.	Ninhydrin	reacts with proline to give :		
	(A)	Purple colour	(B)	Green colour
	(C)	Yellow colour	(D)	Red colour
38.	Conversio	n of two phosphoglycerate to phosph	oenol py	vruvate by the enzyme enolase is:
	(A)	Decarboxylation	(B)	Dehyration
	(C)	Phosphorelation	(D)	Dehydrogenation
39.	Which is	the active form of an enzyme?		
	(A)	Holoenzyme	(B)	
	(C)	Co-enzyme	(D)	Zymogen
40.	The follow	ving are the examples for C18 fatty a	cids EX0	CEPT:
	(A)	Oleic acid	(B)	Palmmitic acid
	(C)	Stearic acid	(D)	Linoleic acid
41.	The biolog	gical indicator used for testing the en	fficiency	of autoclave sterilization is:
	(A)	Geobacillus stearothermophilus	(B)	Bacillus anthracis
	(C)	Bacillus megatarium	(D)	Clostridium botulinum
42.	Amphitri	chous bacteria possess :		
	(A)	One flagellum at one end		
	(B)	Single flagellum at each pole		
	(C)	A cluster of flagella at one end		
	(D)	Uniformly distributed flagella ove	r the wh	ole cell surface

40.	Willen of	the following is the largest at	umai virus:		
	(A)	Rabies virus	(B)	HIV	
	(C)	Pox virus	(D)	Herpes virus	
44.	The first		of bacteria in c	ausing human diseases came from	the
	(A)	Robert Koch	(B)	Louis Pasteur	
	(C)	Joseph Lister	(D)	Edward Jenner	
45.	Which of	the following microscopes can	be used for obs	ervation of living cells?	
	(A)	Bright field Microscope			
	(B)	Dark field Microscope			
	(C)	Phase Contrast Microscope			
	(D)	All of the above			
46.	Which ty	ype of culture media is supp f fastidious microbes?	olemented with	special nutrients for supporting	the
	(A)	Enrichment media	(B)	Enriched media	
	(C)	Selective media	(D)	Differential media	
47.	Which of	the following is a classic mode	el of cellular slin	ne molds?	
	(A)	Dictyostelium	(B)	Myxogastria	
	(C)	Physarium	(D)	Hemitrichia	
48.	Which of	the following is an acidic dye	used in simple s	taining?	
	(A)	Methylene blue	(B)	Basic fuchsin	
	(C)	Crystal violet	(D)	Eosin	
49.	In negative spread with	ve staining of tissue specimen ith a thin film of :	for transmissio	n electron microscopy, the specimen	n is
	(A)	Mercuric chloride	(B)	Platinum	
	(C)	Phosphotungstic acid	(D)	Liquid epoxy plastic	
50.	Bacterial	cell membrane contain steroic	l like molecules	called:	
	(A)	Cholesterol	(B)	Hopanoids	
	(C)	Terpenoids	(D)	Glycerol	

51.	Molds and yeasts which causes food spoilage prefers :				
	(A)	Alkaline pH	(B)	Neutral pH	
	(C)	Acidic pH	(D)	None of the above	
52.		the following selective medium is ter quality analysis?	s used for i	solation of coliforms in the confirmed	
	(A)	Mac Conkey Agar	(B)	Potato Dextrose Agar	
	(C)	Mannitol Salt Agar	(D)	Eosin Methylene Blue Agar	
53.	The most	common causative agent of food b	orne diseas	es in humans is :	
	(A)	Clostridium perfringens	(B)	Clostridium botulinum	
	(C)	Staphylococcus aureus	(D)	Campylobacter jejuni	
54.	Which fur	ngi produces aflatoxins, known to	be powerfu	l carcinogens?	
	(A)	Aspergillus flavus	(B)	Claviceps purpura	
	(C)	Rhizopus stolonifer	(D)	Penicillium notatum	
55.	Which of	the following is a mold ripened ch	eese?		
	(A)	Swiss cheese	(B)	Cottage cheese	
	(C)	Mozzarella	(D)	Blue cheese	
56.	A halophi	ile would be a microorganism that	prefers :		
	(A)	Increased amount of acid	(B)	Increased amount of oxygen	
	(C)	Increased amount of salt	(D)	Increased amount of pressure	
57.	The majo	r biological indicator of faecal poll	ution is:		
	(A)	Escherichia coli	(B)	Streptococcus faecalis	
	(C)	Thiobacillus ferroxidans	(D)	Clostridium perfringens	
58.	The time	-temperature combination of HTS of:	ST pasteuri	zation of 71°C for 15 sec is selected on	
	(A)	Escherichia coli	(B)	Bacillus subtilis	
	(C)	Clostridium botulinum	(D)	Coxiella burnetti	
59.	Which o	f the following organism produ tion of low acid canned foods?	ces nisin,	a bacteriocin currently used for the	
	(A)	Enterococcus faecalis	(B)	Lactococcus lactis	
	(C)	Lactobacillus acidophilus	(D)	Streptococcus lactis	

60.	In which of the following fungal division comes the common bread mold, Rhizopus stolonifer?				
	(A)	Ascomycota	(B)	Zygomycota	
	(C)	Basidiomycota	(D)	Deuteromycota	
61.	Use of liv	ing microorganisms to degrade	environmenta	al pollutants is called :	
	(A)	Micro remediation			
	(B)	Nano remediation			
	(C)	Bio remediation			
	(D)	All of these			
62.	Introduct	ion of DNA in to cells exposing	to high voltag	ge electric pulse is called :	
	(A)	Electrofusion	(B)	Electrofision	
	(C)	Electrolysis	(D)	Electroporation	
63.	The PCR	a technique was developed by :			
	(A)	Kary Mullis	(B)	Kobler	
	(C)	Milstein	(D)	Altman	
64.	Interferor	ns are :			
	(A)	Antibacterial Proteins	(B)	Antiviral Proteins	
	(C)	Bacteriostatic Proteins	(D)	All of these	
65.	The first	successfully cloned animal was			
	(A)	Monkey	(B)	Gibbon	
	(C)	Sheep	(D)	Rabbit	
66.	In E coli,	which enzyme synthesizes the l	RNA primer f	or okazaki fragments :	
	(A)	DNA A	(B)	DNA C	
14	(C)	DNA G	(D)	All of these	
67.	Transcrip	tion is the transfer of genetic in	formation fro	om:	
	(A)	DNA to mRNA	(B)	tRNA to mRNA	
	(C)	tRNA to DNA	(D)	none	
68.	Chlorella	species are widely used in the r	emoval of:		
	(A)	Organic wastes	(B)	Hydrocarbons	
	(C)	Heavy metals	(D)	All of these	

69.	Western blotting is the technique for the detection of:					
	(A)	Specific DNA in a sample	(B)	Specific RNA in a sample		
	(C)	Specific protein in a sample	(D)	Specific glycolypid in a sample		
70.	In which their net		od where p	proteins are separated on the basis of		
	(A)	Affinity chromatography	(B)	Ion Exchange Chromatography		
	(C)	Dialysis	(D)	Gel Filtration Chromatography		
71.	Ti plasmi	d, that is used in plant vector is ob	tained from	m:		
	(A)	Agrobacterium tumifaciens	(B)	Agrobacterium rhizhogenes		
	(C)	Agrobacterium radiobactor	(D)	Thermus aquatics		
72.	Which of	the following reagents are used for	precipitat	ing DNA?		
	(A)	Isopropanol	(B)	Ethanol		
	(C)	Both (A) and (B)	(D)	None of these		
73.	Probiotics	are:				
	(A)	Cancer inducing microbes	(B)	Safe antibiotics		
	(C)	New kind of food allergens	(D)	Live microbial food supplement		
74.	The first	vaccine produced from animal cell	culture wa	as:		
	(A)	Hepatitis B vaccine	(B)	Influenza vaccine		
	(C)	Small pox vaccine	(D)	Polio vaccine		
75.	A group o	of genetically similar organisms obt	tained by a	a sexual reproduction is called:		
	(A)	Clone	(B)	Population .		
	(C)	Assembly	(D)	None		
76.	RFLP is:					
	(A)	(A) Restriction Fragment Length Polymorphism				
	(B)	Repeated Fragment Length Poly	morphism			
	(C)	Renewed Fragment Length Poly				
	(D)	Required Fragment Length Poly	morphism			
				· ·		

	(A)	Occurs after the synthesis has been completed				
	(B)	Is a function of the 3'-5' exonuclease activity of the DNA polymerases				
(C) Requires the presence of an enzyme separate from the DNA polymerases						
	(D)	Occurs in prokaryotes but not	eukaryotes			
78.	Transgen	ic Organisms are :				
	(A)	Produced by gene transfer tech	nology			
	(B)	Extinctict Organisms				
	(C)	Naturally occurring and enden	nic			
	(D)	Produced by traditional plant		nnique		
79.	ECORI is	ė.				
	(A)	DNA Ligase Enzyme				
	(B)	Restriction Endonuclease				
	(C)	A vector used for insulin synth	esis			
	(D)	A plasmid used as a vector				
80.	Monoclon	al antibodies are usually produc	ed from :			
	(A)	Myeloma cells	(B)	Hybridoma cells		
	(C)	Monocytes	(D)	Adipocytes		
81.	The India	n Constitution was amended for	the first tim	ne in :		
	(A)	1950	(B)	1951		
	(C)	1952	(D)	1953		
82.	The Cons Constitut		Fundamental	Duties, from which of these Country	ries'	
	(A)	USA	(B)	Canada		
	(C)	USSR	(D)	UK		
83.	Who acted	d as the Chairman of the Draftin	ng Committe	e of the Constituent Assembly?		
	(A)	B.C. Rajagopalachari	(B)	B.R. Ambedkar		
	(C)	Rajendra Prasad	(D)	Jawaharlal Nehru		
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77. Proof reading activity to maintain the fidelity of DNA synthesis:

84.	When did Right to Information Act come into force?					
	(A)	2005	(B)	2003		
	(C)	2004	(D)	2006		
85.	Who becar	me India's first woman judge at	the Kerala F	ligh Court in 1959?		
	(A)	Arya Pallam	(B)	Akkama Cheriyan		
	(C)	Anna Chandy	(D)	Lalitha Prabu		
86.	Antharjar	na Samaajams were organised t	ınder the lead	lership of:		
	(A)	Akkama Cheriyan	(B)	Lalitha Prabu		
	(C)	Kuttimalu Amma	(D)	Arya Pallam		
87.	Which is	the only Union Territory to hav	e a High Cou	rt of its own?		
	(A)	Chandigarh	(B)	Lakshadweep		
	(C)	Delhi	(D)	Daman and Diu		
88.	Kerala Sa	ngeetha Nataka Academy is lo	cated in :			
	(A)	Ernakulam	(B)	Kottayam		
	(C)	Thrissur	(D)	Kozhikode		
89.	The news	paper started by Herman Gun	dert:			
	(A)	Pashchimodayam	(B)	Rajyasamacharam		
	(C)	Malayalarajyam	(D)	Keralapazhama		
90.	The missi	onary who established the first	printing pres	ss at Kottayam in 1821 :		
	(A)	Joseph Fenn	(B)	Thomas Norton		
	(C)	Henry Baker	(D)	Benjamin Bailey		
91.	'Savarna	Jatha' to support Vaikom Saty:	agraha was oi	ganized by :		
	(A)	K. Kelappan	(B)	Thycad Ayya		
	(C)	Mannathu Padmanaban	(D)	G.P. Pillai		
92.	'Vivekoda	yam' magazine was published	by:			
	(A)	Ramakrishna Pilla	(B)	V.T. Bhattathirippadu		
	(C)	Ayyankali	(D)	Kumaranasan		

(A) 1932 (C) 1909 (D) 1947  94. Who was the founder of Prathyaksha Raksha Daiva Sabha? (A) Ayyankali (B) Poykayil Kumara Guru Devan (C) Sahodaran Ayyappan (D) Vaghbadananda  95. 'Pattini Jatha' was led by: (A) A.K. Gopalan (C) V.T. Bhattathirippadu (D) Ayyankali  96. Which novel of Thakazhi Sivasankara Pillai got Kendra Sahithya Academy Award in 1956 (A) Enippadikal (B) Anubhavangal Palichakal (C) Chemmeen (D) Ouseppinte Makkal  97. Who is known as 'Kerala Vyasa'? (A) Kunjikuttan Thampuran (B) Vallathol Narayana Menon (C) Kumaranasan (D) G. Sankara Kurup  98. Which country adopted its first democratically framed constitution on 20th September 201 (A) Bangladesh (B) Myanmar (C) Bhutan (D) Nepal  99. International day of peace: (A) September 11 (B) September 21 (C) September 24 (D) September 6  100. Which state received the tourism award of 2013-14 for most innovative use of Informat technology? (A) Gujarat (B) Madhya Pradesh (C) Kerala (D) Assam	95.		ation known as Nivarthana n ional reforms of :	novement w	vas started as a protest against th
94. Who was the founder of Prathyaksha Raksha Daiva Sabha?  (A) Ayyankali (B) Poykayil Kumara Guru Devan (C) Sahodaran Ayyappan (D) Vaghbadananda  95. 'Pattini Jatha' was led by: (A) A.K. Gopalan (B) Sree Narayana Guru (C) V.T. Bhattathirippadu (D) Ayyankali  96. Which novel of Thakazhi Sivasankara Pillai got Kendra Sahithya Academy Award in 1956 (A) Enippadikal (B) Anubhavangal Palichakal (C) Chemmeen (D) Ouseppinte Makkal  97. Who is known as 'Kerala Vyasa'? (A) Kunjikuttan Thampuran (B) Vallathol Narayana Menon (C) Kumaranasan (D) G. Sankara Kurup  98. Which country adopted its first democratically framed constitution on 20th September 201 (A) Bangladesh (B) Myanmar (C) Bhutan (D) Nepal  99. International day of peace: (A) September 11 (B) September 21 (C) September 24 (D) September 6  100. Which state received the tourism award of 2013-14 for most innovative use of Informat technology? (A) Gujarat (B) Madhya Pradesh		(A)	1932	(B)	1919
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(C) Sahodaran Ayyappan (D) Vaghbadananda  95. 'Pattini Jatha' was led by:	94.	Who was	the founder of Prathyaksha Rak	sha Daiva S	abha?
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technology?  (A) Gujarat  (B) Madhya Pradesh		(C)	September 24	(D)	September 6
	100.			of 2013-14 f	or most innovative use of Information
(C) Kerala (D) Assam		(A)	Gujarat	(B)	Madhya Pradesh
		(C)	Kerala	(D)	Assam