No. of Printed Pages : 3

**BET-013** 

## DIPLOMA IN CIVIL ENGINEERING DCLE(G) / DIPLOMA IN MECHANICAL ENGINEERING (DME) / DCLEVI / DMEVI / DELVI / DECVI / DCSVI / ACCLEVI / ACMEVI / ACELVI / ACECVI / ACCSVI

28200

## **Term-End Examination**

## December, 2016

## **BET-013 : CHEMISTRY**

Time : 2 hours

Maximum Marks : 70

Note: Attempt five questions in all. Question number 1 is compulsory. Answer any four questions from questions no. 2 to 8. All questions carry equal marks.

1.	Answer	the	follo	wing	questions	in	brief:
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(a)	What is law of octaves ?	2
· (b)	What is a periodic table ? Name the first t elements in the periodic table.	wo 2
(c)	Define metallic character of an element.	2
( <b>d</b> )	Differentiate between an ore and an alloy.	2
(e)	List different types of hardness of water.	2
( <b>f</b> )	What are lubricating oils ?	2
(g)	What is the hydrological cycle ?	2
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2.	(a)	Which one is larger — Na or Na <sup>+</sup> and why ?		
	(b)	Discuss in detail about the modern periodic table.	8	
	(c)	Write the electronic configuration of Cu (29).	2	
<b>3.</b>	(a)	Why is the second ionisation energy higher than the first ionisation energy ?	4	
	(b)	Describe any two methods for preparation of oxygen.	4	
	(c)	How do you extract aluminium from its ores?	6	
4.	(a)	Give the formulae of the following compounds: (i) Dinitrogen monoxide (ii) Nitric oxide (iii) Nitrogen dioxide	4	
		(iv) Dinitrogen pentoxide		
	(b)	Differentiate between specific heat and latent heat.	4	
	(a)	What do you understand by secondary		

(c) What do you understand by secondary fuels ? Give two examples each. 6

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5.	(a)	Mention any two units to represent	
•		hardness of water and explain them.	4
-	<b>(b</b> )	What do you understand by pH ?	2
	(c)	If pOH of a solution is 9, what is its pH ? $(pK_w = 14)$	4
	(d)	Name any four gaseous fuels.	4
<b>6.</b>	(a)	Write a detailed note on semi-solid lubricants including their merits and demerits.	8
	<b>(b</b> )	Explain about viscosity and viscosity index.	6
7.	<b>(a</b> )	What is gutta percha?	2
	(b)	Explain glass transition temperature with a diagram.	6
	(c)	Give an account of the important methods to prevent brittle fracture of glass.	6
8.	(a)	What is the plasticity of clay ?	4
	(b)	How are refractory materials classified ?	6
	(c)	How will you determine actual available chlorine in bleaching powder ?	4

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